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Broglie wavelength and

speed of an electron. The

De-Broglie wavelength is

$= 7.77 \times 10^{-11} \text{ m}$ The

speed of an electron is =

$9.376 \times 10^6 \text{ m s}^{-1}$

Given data: The electron

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potential difference (V)

= 250 V Formula used: $KE = \frac{1}{2} m v^2 = e V$

Where, KE = kinetic

energy m = mass e =

charge of electron v =

velocity V = potential

difference Re-arrange the

equation, $v = \sqrt{\frac{2 e V}{m}}$

1.2 is the required ...

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