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~~Note Taking Guide Episode 701~~ ~~Note Taking Guide: Episode 701 Name: Kelly R. Lab results: 1 doz grains of rice = 0.3 g (Use this fact as a conversion factor.)? grains of rice = 1.94 g~~ ~~Avogadro's Number - the mole = the number 6.022 x 10^23~~ ~~Molar Mass the mass of one mole of any pure substance~~ ~~molar mass of an element~~

~~Note Taking Guide Episode 701 Answers~~

~~CHEMISTRY: A Study of Matter © 2004, GPB 7.1~~ ~~Note Taking Guide: Episode 701 Name~~ ~~Lab results: 1 doz grains of rice = g (Use this fact as a conversion factor.) ? grains of rice = 1.94 g~~ ~~Avogadro's Number - the mole = t he number 6.023x10^23~~ ~~Molar Mass the mass of one mole of any pure substance~~ ~~molar m as s of an element (g/mol) equals the atomic mass of the element in grams~~ ~~molar m as s of a compound equals the sum of the molar mass of the element making up the compound~~ ~~example problems ...~~

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