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The 52nd Colloid and Surface Science Symposium of the Division of Colloid and Surface Chemistry of the American Chemical Society was held in Knoxville, TN, June 12-14, 1978, and one of its Sections was devoted to the topic of Solution Chemistry of Surfactants. Although it was billed as the Section on Solution Chemistry of Surfactants, but it was indeed a veritable international symposium on this topic as 51 papers by about 100 contributors from 12 countries were listed in the program. The present volume and its companion volume 2 document the proceedings of the above-mentioned Section on Solution Chemistry of Surfactants. In 1976 there was held an international symposium on Micellization, Solubilization and Microemulsions in Albany, the proceedings of which have been chronicled in two volumes. A great deal of material dealing with micelles contributed by a legion of prominent researchers constitutes these volumes but a few subtopics were not

adequately covered; so it was deemed appropriate to cover these topics as well as the recent progress in the general area of aggregation of surfactants in this Section. Also as it is the amphiphilicity or amphipathicity* of a surfactant molecule which is responsible for both adsorption at interfaces and aggregation in solution, so it was considered quite apropos to include the topic of adsorption at interfaces in this Section. Concomitantly, the present volumes not only cover the aggregation phenomena but also the adsorption at interfaces.

Stability constants are fundamental to understanding the behavior of metal ions in aqueous solution. Such understanding is important in a wide variety of areas, such as metal ions in biology, biomedical applications, metal ions in the environment, extraction metallurgy, food chemistry, and metal ions in many industrial processes. In spite of this importance, it appears that many inorganic chemists have lost an appreciation for the importance of stability constants, and the thermodynamic aspects of complex formation, with attention focused over the last thirty years on newer areas, such as organometallic chemistry. This book is an attempt to show the richness of chemistry that can be revealed by stability constants, when measured as part of an overall strategy aimed at understanding the complexing properties of a particular ligand or metal ion. Thus, for example, there are numerous crystal structures of the Li^+ ion with crown ethers. What do these indicate to us about the chemistry of Li^+ with crown ethers? In fact, most of these crystal structures are in a sense misleading, in that the Li^+ ion forms no complexes, or at best very weak complexes, with familiar crown ethers such as 12-crown-4, in any known solvent. Thus, without the stability constants, our understanding of the chemistry of a metal ion with any particular ligand must be regarded as incomplete. In this book we attempt to show how stability constants can reveal factors in ligand design which could not readily be deduced from any other physical technique.

Discover the materials set to revolutionize the electronics industry The search for electronic materials that can be cheaply solution-processed into films, while simultaneously providing quality device characteristics, represents a major challenge for materials scientists. Continuous semiconducting thin films with large carrier mobilities are particularly desirable for high-speed microelectronic applications, potentially providing new opportunities for the development of low-cost, large-area, flexible computing devices, displays, sensors, and solar cells. To date, the majority of solution-processing research has focused on molecular and polymeric organic films. In contrast, this book reviews recent achievements in the search for solution-processed inorganic semiconductors and other critical electronic components. These components offer the potential for better performance and more robust thermal and mechanical stability than comparable organic-based systems. Solution Processing of Inorganic Materials covers everything from the more traditional fields of sol-gel processing and chemical bath deposition to the cutting-edge use of nanomaterials in thin-film deposition. In particular, the book focuses on materials and techniques that are compatible with high-throughput, low-cost, and low-temperature deposition processes such as spin coating, dip coating, printing, and stamping. Throughout the text, illustrations and examples of applications are provided to help the reader fully appreciate the concepts and opportunities involved in this exciting field. In addition to presenting the state-of-the-art research, the book offers extensive background material. As a result, any

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researcher involved or interested in electronic device fabrication can turn to this book to become fully versed in the solution-processed inorganic materials that are set to revolutionize the electronics industry.

Advanced Organic Chemistry: Reactions and Mechanisms covers the four types of reactions — substitution, addition, elimination and rearrangement; the three types of reagents — nucleophiles, electrophiles and radicals; and the two effects — electroni

This outline of the principles and chemical interactions in inorganic solution chemistry delivers a course module in an area of considerable complexity. Problems with solutions and tutorial hints to test comprehension have been added as a feature to check readers' understanding and assist self-study. Exercises and projects are also provided to help readers deepen and extend their knowledge and understanding. Inorganic solution chemistry is treated thoroughly Emphasis is placed upon NMR, UV-VIS, IR Raman spectroscopy, X-ray diffraction, and such topics as acid-base behaviour, stability constants and kinetics

Providing a comprehensive review of the state-of-the-art advanced research in the field, Polymer Physics explores the interrelationships among polymer structure, morphology, and physical and mechanical behavior. Featuring contributions from renowned experts, the book covers the basics of important areas in polymer physics while projecting into the future, making it a valuable resource for students and chemists, chemical engineers, materials scientists, and polymer scientists as well as professionals in related industries.

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