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*Physics Fix 27 Information: Waves and Jumping Rope*

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10. Using only your answers to questions 7-9, calculate how many Joules of energy it takes to change 12.7 g of water at 22°C to steam at 159°C. Questions 7-9 are each different "steps" for taking 12.7g of water from 22°C to 159°C so you can simply add the answers:  $4140.7 + 28,702 + 1513.6 = 34,356.3$  J. 11.

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*Solved: Chapter 27, Problem 075 Flying Circus Of Physics B*

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Let  $\vec{v}_1 = -6.3 \hat{i} + 8.0 \hat{j}$  and  $\vec{v}_2 = -4.9 \hat{j} + 4.3 \hat{i}$ . A) Determine the magnitude of  $\vec{v}_1$ . B) Determine the direction of  $\vec{v}_1$ . C) Determine the magnitude of  $\vec{v}_2$ . C) Determine the direction of  $\vec{v}_2$ .

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