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The solution contains 100 meq of the compound (conjugate base plus acid), so 80 meq are in the conjugate base form

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and 20 meq are in the acid form, for a [base]/ [acid] ratio of 4. Because HCl is a...

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Water Explain why ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ ) is more soluble in  
water than is ethane ( $\text{CH}_3\text{CH}_3$ ). Answer Ethanol is polar;

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ethane is not.

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